

DETECTION OF ADULTERATION IN DAIRY PRODUCTS: A REVIEW

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Abstract

Milk and dairy products are among the most nutritious foods consumed worldwide because they provide proteins, carbohydrates, fats, vitamins, and minerals essential for human health. However, increasing demand and commercial competition have led to the widespread adulteration of dairy products. Adulterants such as water, starch, urea, detergents, glucose, formaldehyde, and synthetic milk reduce nutritional quality and may cause serious health problems, including gastrointestinal disorders, kidney damage, metabolic imbalance, and long-term toxic effects. The detection of these adulterants is therefore essential for ensuring food safety and maintaining consumer confidence. Various physical, chemical, and instrumental techniques are available to detect these adulterants. This review summarizes the major adulterants commonly found in milk and dairy products, their adverse health effects, laboratory detection methods, and the importance of regulatory quality control, consumer awareness, and routine monitoring for ensuring the safety and authenticity of dairy products.

Keywords

Dairy products, Milk adulteration, Food safety, Chemical tests, Quality control

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Introduction

Milk is considered one of the most complete natural foods because it contains almost all essential nutrients required for growth and development. It is an important source of proteins, calcium, lactose, vitamins, and minerals. Dairy products such as curd, butter, cheese, paneer, ghee, and milk powder are widely consumed throughout the world.

The increasing demand for dairy products has also increased the incidence of adulteration. Adulteration involves the intentional addition of inferior or harmful substances to increase volume, improve appearance, or extend shelf life. Such practices not only reduce the nutritional value of milk but also pose significant health risks. Therefore, the detection of adulteration has become an important aspect of food quality assurance.

Dairy products occupy a significant place in the human diet because they provide high-quality proteins, essential amino acids, calcium, phosphorus, vitamins A, D and B-complex, and other bioactive compounds. These nutrients contribute to bone development, muscle growth, immune function, and overall health. Due to their high nutritional value and widespread consumption across all age groups, maintaining the quality and safety of dairy products is of utmost importance.

In recent years, globalization of the dairy industry and increasing market competition have increased the risk of economically motivated adulteration. Unscrupulous manufacturers may intentionally add cheaper substances to maximize profits while compromising product quality. Such adulteration not only deceives consumers but also creates serious public health concerns. Therefore, rapid, reliable, and accurate methods for the detection of adulterants are essential for ensuring food safety and regulatory compliance.

Common Adulterants in Dairy Products

Several substances are commonly added to milk and dairy products for economic gain. Some of the major adulterants include:

- **Water:** Added to increase milk volume, resulting in reduced nutritional value.
- **Starch:** Used to increase the apparent density of diluted milk.
- **Urea:** Added to artificially increase protein content.
- **Detergents:** Used in synthetic milk to improve foaming and emulsification.
- **Glucose and Sucrose:** Added to restore sweetness after dilution.
- **Synthetic Milk:** Prepared using detergents, oils, urea, and other chemicals to imitate natural milk.
- **Formaldehyde:** Used as a preservative to delay spoilage.

- **Salicylic Acid and Benzoic Acid:** Added to prolong shelf life.
- **Melamine:** Increases the apparent nitrogen content during protein analysis.
- **Neutralizers:** Sodium carbonate or sodium bicarbonate are added to neutralize acidity in spoiled milk.

Different adulterants are added for different economic purposes. Water is commonly used to increase the volume of milk, while starch, glucose, and sucrose help restore its consistency and taste after dilution. Urea and melamine artificially increase the apparent protein content during quality testing. Preservatives such as formaldehyde and salicylic acid delay microbial spoilage, allowing adulterated milk to remain marketable for longer periods. Synthetic milk, prepared using detergents, refined oils, urea, and emulsifiers, closely resembles natural milk in appearance but lacks its nutritional value and may be hazardous to human health.

Adulteration may occur at different stages of the dairy supply chain, including milk collection, transportation, storage, processing, and retail distribution. Inadequate quality control and weak regulatory monitoring further increase the chances of adulterated dairy products reaching consumers. Therefore, strict surveillance and regular testing are necessary to maintain product authenticity. These adulterants compromise food quality and may produce both short-term and long-term health hazards.

Factors Responsible for Milk Adulteration

Milk adulteration is influenced by several economic, technical, and regulatory factors. The major factors responsible are:

Economic Profit: The primary reason for milk adulteration is to increase profit by adding cheaper substances such as water, starch, urea, or detergents. This practice increases the quantity of milk while reducing production costs.

Increasing Consumer Demand: Rapid population growth and the rising demand for milk and dairy products often exceed production capacity. This encourages some producers to adulterate milk to meet market demand.

Poor Quality Control: Inadequate monitoring during milk collection, transportation, storage, and processing increases the possibility of adulteration. Lack of routine quality testing allows adulterated milk to enter the market.

Seasonal Variation in Milk Production: Milk production decreases during certain seasons due to environmental conditions and reduced animal productivity. To compensate for the shortage, adulterants may be added to maintain supply.

Weak Regulatory Enforcement: Insufficient implementation of food safety regulations, limited laboratory facilities, and irregular inspections contribute to the continued practice of milk adulteration in some regions.

Lack of Awareness: Some small-scale dairy farmers and vendors have limited knowledge of food safety standards and the harmful effects of adulteration. Similarly, consumers may be unaware of methods to identify adulterated milk.

Poor Storage and Transportation: Improper handling and transportation of milk may lead to spoilage. To delay spoilage or improve the appearance of deteriorated milk, preservatives or neutralizers are sometimes added illegally.

Intense Market Competition: High competition among dairy producers and suppliers may encourage unethical practices to reduce production costs and maximize profits, ultimately compromising milk quality and consumer safety.

Recent Advances in Dairy Adulteration Detection

In addition to conventional physical, chemical, and instrumental methods, several advanced technologies have been developed for the rapid, accurate, and sensitive detection of adulterants in milk and dairy products. These modern approaches improve analytical efficiency, reduce testing time, and provide reliable results even at very low concentrations of adulterants.

Biosensors: Biosensors are highly sensitive analytical devices that combine biological recognition elements with electronic detectors. They enable rapid and accurate detection of adulterants such as urea, glucose, antibiotics, and melamine in milk with minimal sample preparation.

Nanotechnology-Based Sensors: Nanomaterials such as gold nanoparticles, silver nanoparticles, and carbon nanotubes enhance the sensitivity and selectivity of milk analysis. These nanosensors can detect adulterants at trace levels, making them valuable for routine quality assessment.

Immunoassay Techniques: Immunological methods, including Enzyme-Linked Immunosorbent Assay (ELISA), utilize antigen–antibody interactions for the detection of specific contaminants and adulterants. These techniques are highly specific and widely used in food safety laboratories.

Portable Milk Analyzers: Modern portable milk analyzers provide rapid on-site evaluation of milk quality by measuring parameters such as fat, protein, lactose, density, and added water. These instruments are increasingly used by dairy industries and milk collection centers.

Artificial Intelligence (AI) and Digital Technologies: Artificial intelligence and machine learning are emerging tools for dairy quality monitoring. AI-based systems can analyze large datasets, identify abnormal patterns, and improve the accuracy of adulteration detection, supporting efficient quality control and regulatory surveillance.

Methods for Detection of Milk Adulteration

Different analytical techniques are employed depending on the type of adulterant.

1. Physical Methods: Physical examination includes determination of density, viscosity, freezing point, colour, taste, odour, and specific gravity. These methods are simple and suitable for preliminary screening. Physical methods are simple, rapid, and inexpensive techniques that provide preliminary information regarding milk quality. However, they cannot identify specific adulterants with high accuracy and are therefore generally used as screening methods before detailed laboratory analysis.

2. Chemical Methods: Chemical tests remain the most widely used laboratory methods because they are economical, rapid, and reliable. Colour changes or precipitate formation indicate the presence of specific adulterants. Chemical methods are widely preferred in routine laboratories because they require minimal instrumentation and provide rapid results. Most of these methods are based on colour reactions, precipitation reactions, or oxidation–reduction reactions. These tests are suitable for the qualitative detection of common adulterants and are extensively employed in dairy quality control laboratories.

3. Instrumental Methods: Modern analytical instruments such as FTIR spectroscopy, HPLC, GC-MS, UV-Visible spectrophotometry, and chromatography provide accurate identification and quantification of adulterants with high sensitivity. Advanced analytical techniques offer greater sensitivity, specificity, and accuracy than conventional methods. Spectroscopic and chromatographic techniques can simultaneously identify multiple adulterants even at very low concentrations. Although these methods require sophisticated instruments and trained personnel, they provide reliable results for regulatory testing and research purposes.

Laboratory Detection of Common Adulterants

Various qualitative tests are routinely used in dairy laboratories:

Adulterant	Detection Method / Reaction
Detergent	Methylene Blue Test
Urea	DMAB (p-Dimethylaminobenzaldehyde) Test
Ammonium Salts	Nessler's Reagent
Glucose	Barfoed Test
Starch	Iodine Test (produces blue colour)
Sodium Chloride	Silver Nitrate Test (produces white precipitate)
Nitrate	Diphenylamine Test (produces blue colour)
Sulphate	Barium Chloride Test (produces white precipitate)
Formaldehyde	Leach Test
Neutralizers	Phenol Test

Laboratory detection methods play a vital role in ensuring the authenticity and safety of milk. Most qualitative tests are based on characteristic chemical reactions between the adulterant and specific reagents, producing a distinct colour change or precipitate. Positive results obtained through these screening methods are often confirmed using advanced instrumental techniques. Regular laboratory analysis helps regulatory agencies detect fraudulent practices and protect consumer health.

The selection of an appropriate detection method depends on factors such as the type of adulterant, sample composition, analytical sensitivity, cost, and available laboratory facilities. Combining conventional chemical tests with modern instrumental techniques provides the most reliable assessment of milk quality. These laboratory methods are inexpensive, easy to perform, and suitable for routine quality control.

Health Effects of Dairy Adulterants

Consumption of adulterated milk can adversely affect human health.

- **Water contamination:** May introduce pathogenic microorganisms.
- **Urea:** May damage kidneys and disturb metabolism.
- **Detergents:** Can cause gastrointestinal irritation and food poisoning.
- **Formaldehyde:** Is toxic and prolonged exposure may increase the risk of cancer.
- **Melamine:** Can lead to kidney stone formation and renal failure.
- **Excessive nitrate and sulphate:** May cause digestive disorders and other systemic complications.

Children, pregnant women, and elderly individuals are particularly vulnerable to the harmful effects of adulterated dairy products.

Continuous consumption of adulterated dairy products may result in both acute and chronic health complications. Mild symptoms include nausea, vomiting, abdominal pain, and diarrhoea, whereas prolonged exposure may damage vital organs such as the kidneys, liver, and nervous system. Children are particularly susceptible because they consume relatively larger amounts of milk during growth and development.

Certain adulterants such as melamine and formaldehyde have been associated with severe toxicological effects. Long-term exposure may increase the risk of renal disorders, carcinogenicity, metabolic imbalance, and immune dysfunction. Therefore,

preventing adulteration is essential not only for maintaining food quality but also for safeguarding public health.

Quality Compliance and Consumer Awareness

Regular monitoring of dairy products according to food safety standards is essential. Good manufacturing practices (GMP), hygienic processing, routine laboratory testing, and strict regulatory surveillance help reduce adulteration. Consumer awareness regarding certified dairy products and proper quality testing also plays an important role in ensuring food safety.

In India, the Food Safety and Standards Authority of India (FSSAI) establishes quality standards and regulations for milk and dairy products. Compliance with these standards requires regular inspection, laboratory testing, proper labeling, hygienic processing, and traceability throughout the production chain. Dairy industries are also encouraged to implement Hazard Analysis and Critical Control Point (HACCP) systems and Good Manufacturing Practices (GMP) to minimize the risk of contamination and adulteration.

Public awareness campaigns, educational programs, and consumer participation can significantly reduce adulteration by encouraging the purchase of certified dairy products from reliable sources. Portable testing kits and digital quality monitoring systems have further strengthened the detection and prevention of milk adulteration.

Conclusion

Milk adulteration remains a significant challenge for food safety and public health. The addition of substances such as water, starch, urea, detergents, formaldehyde, and synthetic chemicals decreases nutritional quality and may lead to serious health complications. Physical, chemical, and instrumental methods provide effective means for detecting adulterants in dairy products. Regular quality assessment, strict regulatory control, and increased consumer awareness are essential to ensure the availability of safe and high-quality dairy products.

Future advancements in analytical technologies, including biosensors, nanotechnology-based sensors, artificial intelligence-assisted quality monitoring, and portable detection devices, are expected to improve the speed and accuracy of adulteration detection. Continuous collaboration among researchers, regulatory authorities, dairy industries, and consumers will be essential for ensuring the production and distribution of safe, nutritious, and authentic dairy products.

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